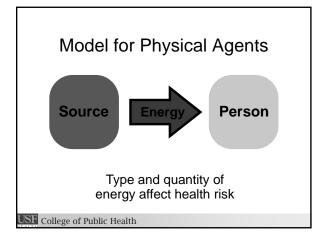
Physical Agents in Environmental and Occupational Health

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Some Types

Radiation

- ✓ Ionizing
- ✓ Nonionizing

Noise

Thermal: Heat and Cold

Vibration Plus

- ✓ Electricity
- ✓ Impact and Shock Waves
- ✓ Pressure (Diving and Altitude)

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Quantity

Total Amount of Energy Absorbed (Work)

- √ What does it take to raise water temperature?
- ✓ Joules

Rate of Absorption (Power)

- ✓ How fast does the temperature rise?
- √ Watts (J/s)

Normalized to Surface Area (W/m², mW/cm²)

Frame of Reference

✓ Basal Metabolic Rate is 6 mW/cm²

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Bernard Watt-O-Meter



The BWOM does not exist in this reality; it may violate known physical laws and grossly simplifies others.

But perhaps it will give you some perspective.

If you care to know how I came up with a number, ask.

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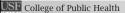
Ionizing Radiation

USF SOUTH FLORID

Q1

Who discovered X-rays? Hint: First example was radiograph of wife's hand.

- a. Becquerel
- b. Curie
- c. Roentgen
- d. Seivert
- e. Zen





Q2

What disease is most associated with low-level exposures to ionizing radiation? Hint: By 1940s, incidence among physicians, especially radiologists, was higher than general population.

- a. leukemia
- b. lung cancer
- c. schizophrenia
- d. tuberculosis
- e. xeroderma pigmentosum

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Q3

Among consumer products, what is the greatest source of ionizing radiation exposure?

- a. Cigarettes
- b. Gas stoves
- c. Old luminous watch dials (pre-60s)
- d. Smoke detectors
- e. Televisions

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What happened 113 years ago?

In October 1895, Wilhelm Roentgen Discovered X-rays

Packets of Energy Called Photons

 $\checkmark \epsilon$ = 12 eV to 10⁸ eV

√ Can Ionize Atoms



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Results of Ionization

Break DNA -- Direct Hit

Create Free Radicals and Peroxide

Results on Cell Viability

- ✓ No Effect
- ✓ Cell Dies
- √Takes a Step on Carcinogenic Pathway

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Biological Effects

Determinants of Biological Effects

- ✓ Rate of absorption
- ✓ Total dose
- ✓ Tissue exposed
- ✓ Individual variations

Classes of Effects

- √ Acute somatic effects (acute radiation sickness)
- ✓ Delayed somatic effects (leukemia, cancers)
- √ Genetic effects (birth defects)

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Energy Levels

 $LD_{50} = 0.3 \text{ mW/cm}^2$

One Time = 0.01 mW/cm²

Occupational = 0.00000003 mW/cm² over 1 yr

Public = 0.0000000008 mW/cm² over 1 yr

70 kg person; gamma / x-ray only

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Types

Electromagnetic Radiation / Photons

- √ X-rays
- √Gamma Rays

Particle Radiation

- ✓ Alpha
- ✓ Beta
- ✓ Neutron

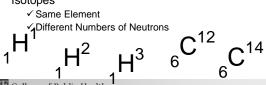
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Chemical Elements

Element

- ✓ Defined by Atomic Number
- ✓ Atomic Number Equals Number of Protons

Isotopes



Nuclear Radiation

Nature Seeks Stability

Radioisotopes

- ✓ Instability occurs when the right blend is not present
- ✓ Nature attempts to create the right blend by radioactive decay.

Nuclear Radiation is Result

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Types of Nuclear Radiation

Alpha Particles

- √ Helium nucleus (2 protons + 2 neutrons)
- ✓ Positive charge (+2)
- ✓ Dissipate energy quickly
- √Travel short distances
- √ Stopped by sheet of paper / skin

 $\{_{2}He^{4^{+2}}\}$

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Types of Nuclear Radiation

Alpha Particles

 $\{e^-\}$

Beta Particles

- ✓ Electron (neutron to proton)
- ✓ Negative charge (-1)
- ✓ Loss energy over short distance
- √ Stopped by aluminum foil / skin (deeper)

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Decay Series

92U238
$$\alpha$$
90Th234 β {e⁻}
91Pa234 β
92U234 α {2He4+2} 92U234
90Th230 α
88Ra226 α
86Rn222

Types of Nuclear Radiation

Alpha Particles Beta Particles

Gamma Rays

- √ Excess energy dissipation from nucleus
- ✓ Photons with high energy
- √ Travel great distances
- ✓ Give up energy slowly
- ✓ Stopped by lead, concrete

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Q4

What distinguishes the energy paths associated with alpha and gamma radiation?

- a. Gamma radiation is readily stopped.
- b. Alpha radiation passes through lead more easily than gamma radiation.
- c. An alpha radiation source is a problem only when it is in the body while a gamma source can be a problem inside or outside.

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Measurement Units

Activity (Ci / Beq)

Radiation Absorbed Dose (rad / Gray)

rad Equivalent - Man (rem / Sievert)

- ✓ Relative Biological Effectiveness (RBE)
- ✓ Gamma has least effect (RBE = 1)
- ✓ Alpha has greatest effect (RBE = 20)
- ✓ Beta tends to be closer to Gamma (RBE = 1 to 5)

rem = RBE x rad

 $Sv = RBE \times Gr$

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Half-Life Time to reduce activity by 1/2

4 half-lives is a 95% reduction.

Short half-life is related to high activity.

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Q5

For the same amount of potential damage (risk) to a particular organ

 $\ensuremath{\mathsf{A}}.$ alpha particles cannot cause any problems, but beta particles can.

B. the amount of energy deposited in the organ is the same, no matter what the type of radiation is.

 $\ensuremath{\text{C}}.$ gamma rays require more total energy than alpha particles.

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Radiation Threat

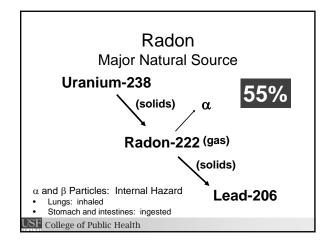
Sources

- ✓ Natural
- ✓ Artificial

Hazards

- ✓ External (long distances)
 - Gamma rays
 - X-rays
- ✓ Internal (short distances)
 - Alpha
 - Beta

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Other Natural Sources

Internal: 11%

Cosmic: 8%

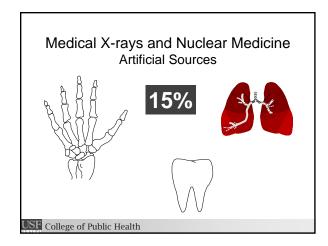
27%

Terrestrial: 8%

Humans may enhance exposure to natural

sources.

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Other Artificial Sources

Consumer Products: 3%

Occupational: 0.3%

3.4%

Nuclear Fuel Cycle: 0.1%

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Commercial Power Generation

Nuclear Fuel Cycle: 0.1%

- ✓ Mining and Milling
- ✓ Enrichment and Fuel Fabrication
 - Japan Incident
- ✓ Power Generation
 - Routine and Catastrophic Releases
 - Three Mile Island
 - Chernobyl
- √Waste Disposal

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Waste Disposal

High-Level Wastes

- ✓ Nuclear Waste Policy Act (1982)
 - Federal Responsibility
- Permanent: Yucca Mountain (Open in 20??)
 Temporary: Monitored Retrievable Storage
- √ Military (weapons) Waste: New Mexico

Low-Level Wastes

- ✓ Low-Level Waste Policy Act (1980)

 - State Responsibility
 Groups of States form Compacts
- √ Siting Problems

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Q6

What is the role of public health professionals in trying to influence individual decisions concerning ionizing radiation?

- A. Ignore it
- Prudent avoidance B.
- C. As low as reasonably achievable (ALARA)
- D. Education on risks and actions

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